

Appendix A

Supplementary data

for

The state of platinum in pyrrhotite: X-ray absorption spectroscopy study and implications for the role of Fe sulfides as platinum carriers

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Table A.1. The operating conditions of EPMA.

EPMA conditions	Pt	Fe	S
Crystal analyser	PETH	LIF	PETH
Analytical line	PtM α	K α	K α
Accelerating voltage	20 kV		
Current in the Faraday cup	20 nA		
Exposure	30 s	20 s	10 s
Probe diameter	1 μ m		
Detection limit (2 σ)	600 ppm	270 ppm	130 ppm

Table A.2. The operating conditions of LA-ICP-MS.

LA-ICP-MS conditions	
RF power	1350 W
Plasma gas	14.2 L min ⁻¹ Ar
Auxiliary gas	0.7 L min ⁻¹ Ar
Make-up gas	0.8 L min ⁻¹ Ar
Sampling depth	1.5 mm
Detector	Dual (pulse and analogue counting)
Dwell time/mass	10 ms
Laser parameters	
Wavelength	213 nm
Energy density	4-7 J cm ⁻²
Carrier gas	0.6 He
Ablation style	Single spot/ line
Ablation spot size	60/40 μ m
Repetition rate	10 Hz

Table A.3. Composition of Pt-bearing pyrite and pyrrhotite measured by EPMA and LA-ICP-MS.

WDS Sample (Point) id	Fe, wt. %	S, wt. %	Pt, wt. %	Fe, at. %	S, at. %	Pt, at. %
Synthesis temperature $t = 720\text{ }^{\circ}\text{C}$						
pyrrhotite, 5602 (1)	59.74	38.97	0.57	46.75	53.12	0.13
pyrrhotite, 5602 (2)	59.60	39.13	0.60	46.59	53.28	0.13
pyrrhotite, 5602 (3)	59.35	39.17	0.58	46.46	53.41	0.13
pyrrhotite, 5602 (4)	59.44	39.25	0.61	46.45	53.42	0.14
pyrrhotite, 5602 (5)	59.44	39.37	0.59	46.37	53.50	0.13
pyrrhotite, 5602 (6)	59.14	38.92	0.58	46.53	53.34	0.13
pyrrhotite, 5602 (7)	59.83	39.06	0.53	46.74	53.14	0.12
pyrrhotite, 5602 (8)	59.96	38.87	0.47	46.92	52.98	0.11
pyrrhotite, 5602 (9)	60.12	38.84	0.36	47.02	52.90	0.08
Average content	59.6	39.1	0.5	46.7	53.2	0.12
$\pm 2\text{SD}$	0.6	0.4	0.2	0.4	0.4	0.04
Synthesis temperature $t = 650\text{ }^{\circ}\text{C}$						
pyrrhotite, 5601 (1)	59.87	38.58	0.34	47.08	52.84	0.08
pyrrhotite, 5601 (2)	60.54	38.57	0.31	47.37	52.56	0.07
pyrrhotite, 5601 (3)	60.05	38.65	0.33	47.11	52.81	0.07
pyrrhotite, 5601 (4)	60.19	38.68	0.31	47.15	52.78	0.07
pyrrhotite, 5601 (5)	60.13	38.82	0.30	47.04	52.89	0.07
pyrrhotite, 5601 (6)	60.03	38.67	0.30	47.09	52.84	0.07
pyrrhotite, 5601 (7)	60.62	39.12	0.07	47.07	52.91	0.02
Average content	60.2	38.7	0.3	47.1	52.8	0.06
$\pm 2\text{SD}$	0.6	0.4	0.2	0.2	0.2	0.04
pyrite, 5601 (1)	46.45	52.13	0.19	33.83	66.13	0.04
pyrite, 5601 (2)	45.95	52.23	1.00	33.49	66.30	0.21
pyrite, 5601 (3)	46.08	51.73	0.29	33.82	66.12	0.06
pyrite, 5601 (4)	45.75	51.46	1.01	33.72	66.06	0.21
Average content	46.1	51.9	0.6	33.7	66.2	0.13
$\pm 2\text{SD}$	0.6	0.7	0.9	0.4	0.9	0.19
pyrrhotite, 5599 (1)	61.46	37.13	0.017 ^A	48.73	51.27	0.004
pyrrhotite, 5599 (2)	61.65	37.23	0.017 ^A	48.74	51.26	0.004
pyrrhotite, 5599 (3)	61.41	37.22	0.017 ^A	48.64	51.35	0.004
pyrrhotite, 5599 (4)	62.05	37.34	0.017 ^A	48.82	51.17	0.004
pyrrhotite, 5599 (5)	61.43	37.11	0.017 ^A	48.73	51.27	0.004
pyrrhotite, 5599 (6)	61.74	37.09	0.017 ^A	48.87	51.13	0.004
Average content	61.6	37.2	0.02^A	48.8	51.2	0.004
$\pm 2\text{SD}$	0.5	0.2	<0.01^A	0.2	0.2	<0.001
pyrrhotite, 5592 (1)	60.17	38.73	0.32	47.11	52.82	0.07
pyrrhotite, 5592 (2)	60.09	38.9	0.36	46.97	52.95	0.08
pyrrhotite, 5592 (3)	59.78	38.63	0.31	47.01	52.92	0.07
pyrrhotite, 5592 (4)	59.89	38.86	0.28	46.92	53.02	0.06
pyrrhotite, 5592 (5)	59.99	38.89	0.28	46.94	53.00	0.06
pyrrhotite, 5592 (6)	60.19	38.67	0.11	47.18	52.79	0.02
Average content	60.0	38.8	0.3	47.0	52.9	0.06
$\pm 2\text{SD}$	0.3	0.2	0.2	0.2	0.2	0.04

Table A.3. – continued.

WDS Sample (Point) id	Fe, wt. %	S, wt. %	Pt, wt. %	Fe, at. %	S, at. %	Pt, at. %
pyrite, 5592 (1)	43.33	51.67	1.59	32.39	67.27	0.34
pyrite, 5592 (2)	46.36	51.90	0.45	33.87	66.04	0.09
pyrite, 5592 (3)	45.19	51.39	0.61	33.51	66.36	0.13
pyrite, 5592 (4)	45.94	51.93	0.39	33.66	66.26	0.08
pyrite, 5592 (5)	45.32	51.78	2.47	33.27	66.21	0.52
pyrite, 5592 (6)	46.42	52.14	0.17	33.81	66.15	0.04
Average content	45.4	51.8	1.0	33.4	66.4	0.20
±2SD	2.3	0.5	1.8	1.7	0.7	0.38
pyrrhotite, 5590 (1)	60.09	38.72	0.31	47.09	52.84	0.07
pyrrhotite, 5590 (2)	59.22	38.67	0.38	46.75	53.17	0.09
pyrrhotite, 5590 (3)	59.78	38.78	0.37	46.91	53.00	0.08
pyrrhotite, 5590 (4)	59.83	38.78	0.38	46.93	52.98	0.09
pyrrhotite, 5590 (5)	59.65	38.71	0.32	46.91	53.02	0.07
pyrrhotite, 5590 (6)	59.45	38.5	0.37	46.95	52.96	0.08
pyrrhotite, 5590 (7)	59.88	38.57	0.37	47.09	52.83	0.08
Average content	59.7	38.7	0.4	47.0	53.0	0.08
±2SD	0.6	0.2	0.1	0.2	0.2	0.02
pyrite, 5590 (1)	45.88	52.96	0.40	33.19	66.73	0.08
pyrite, 5590 (2)	45.05	51.99	2.12	33.07	66.48	0.45
pyrite, 5590 (3)	46.03	52.09	0.33	33.64	66.30	0.07
pyrite, 5590 (4)	45.65	52.16	1.67	33.33	66.32	0.35
pyrite, 5590 (5)	45.63	52.22	1.02	33.34	66.45	0.21
pyrite, 5590 (6)	44.97	51.60	1.88	33.22	66.39	0.40
Average content	45.5	52.2	1.2	33.3	66.4	0.26
±2SD	0.9	0.9	1.5	0.6	1.1	0.32

^A determined via LA-ICP-MS analysis

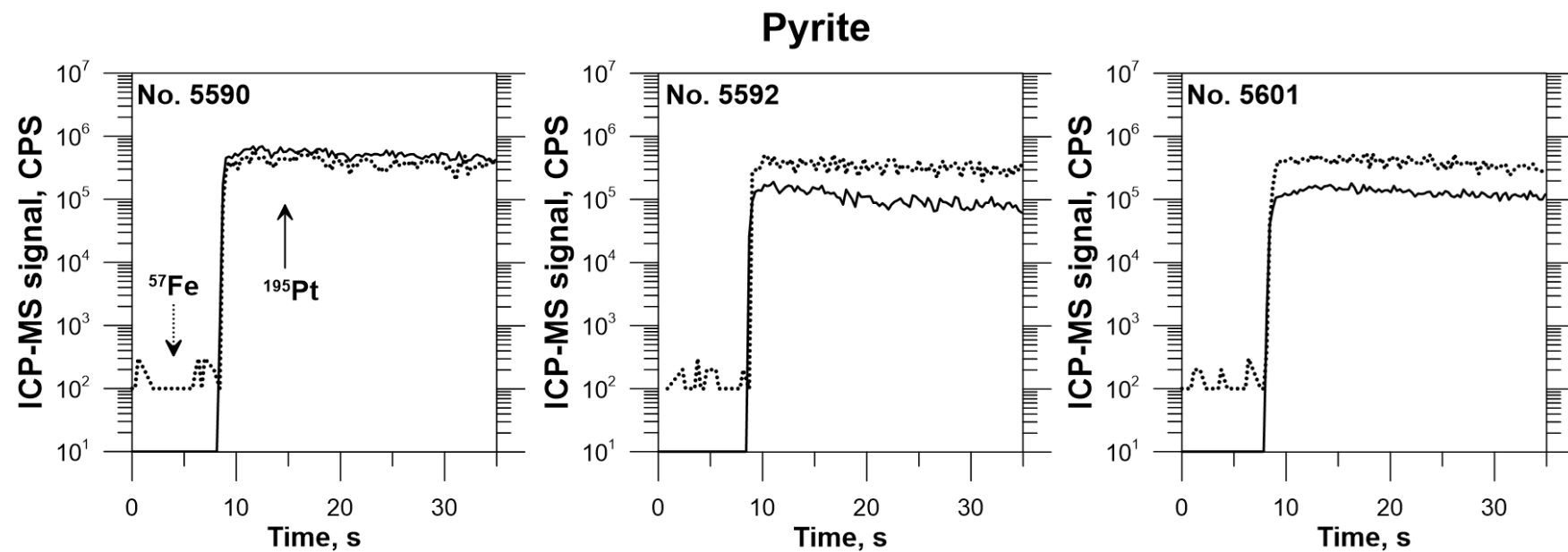


Figure A.1. Time-resolved laser ablation ICP-MS spectra of Pt-bearing synthetic pyrites. Analyses were performed using a spot size of 60 μm . The ^{195}Pt (solid lines) disseminated in the pyrite matrix behaves similarly to the internal standard element – ^{57}Fe (dotted lines).

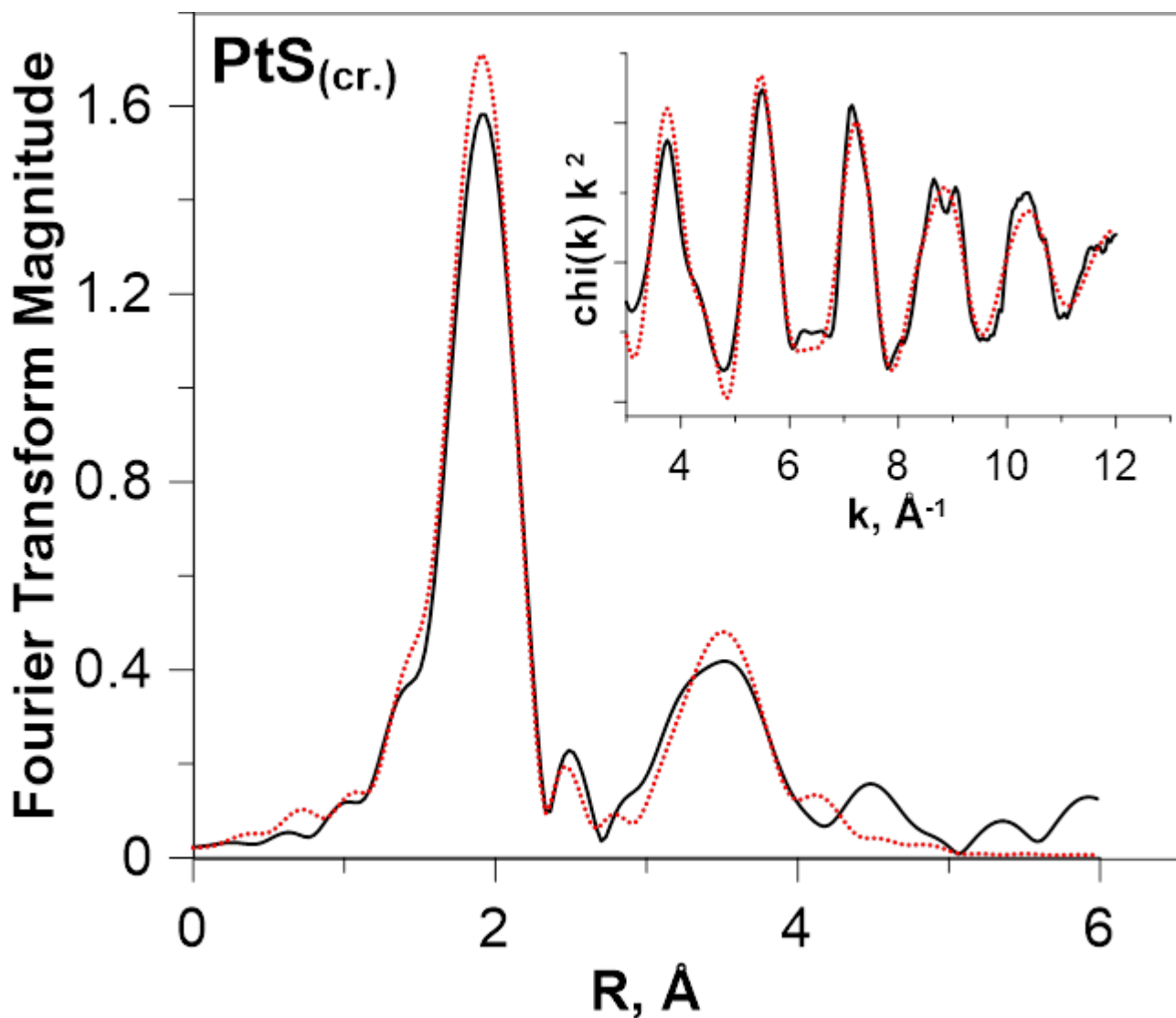
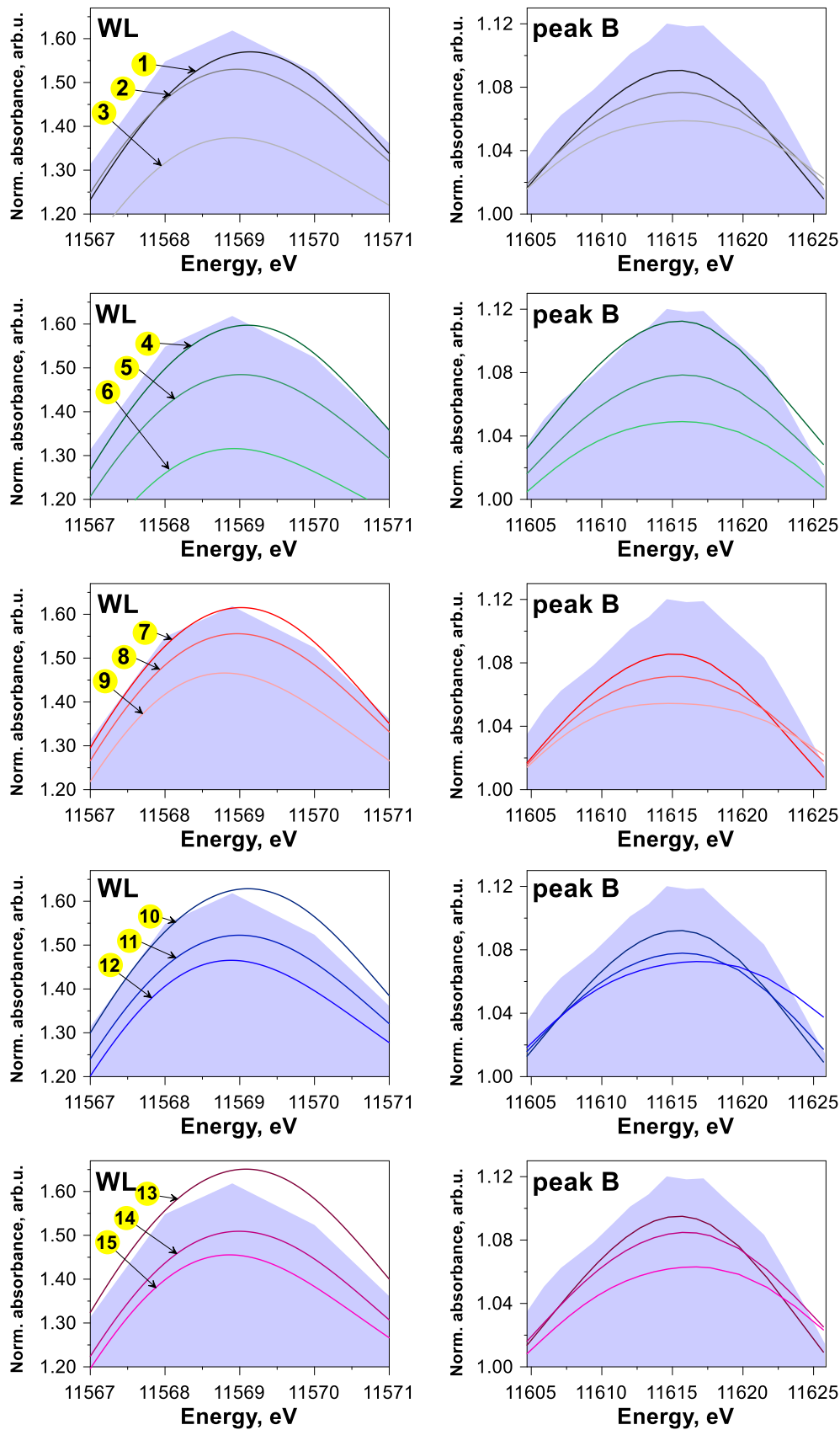
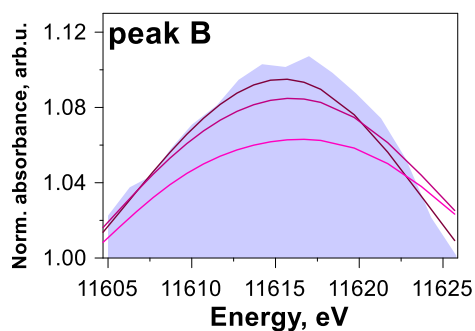
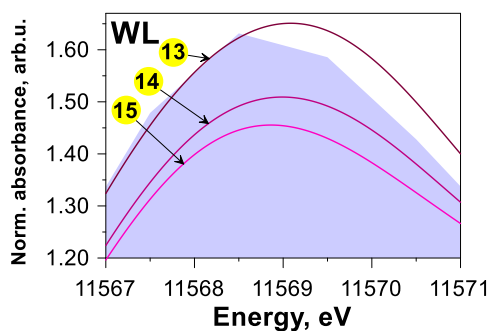
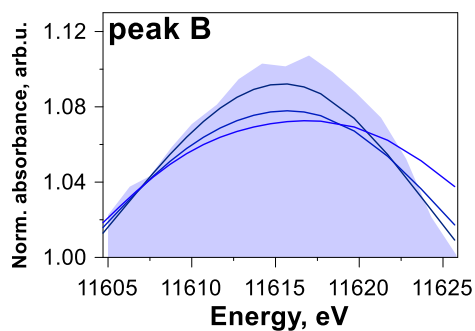
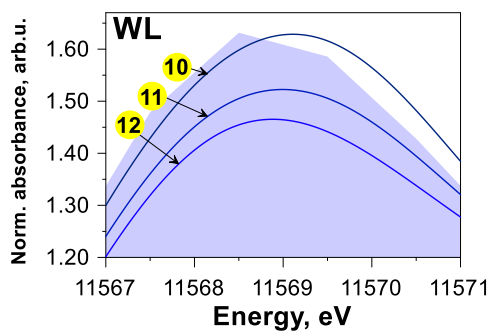
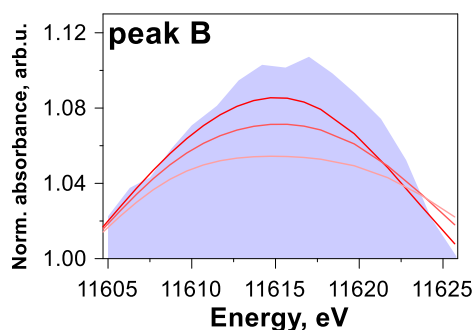
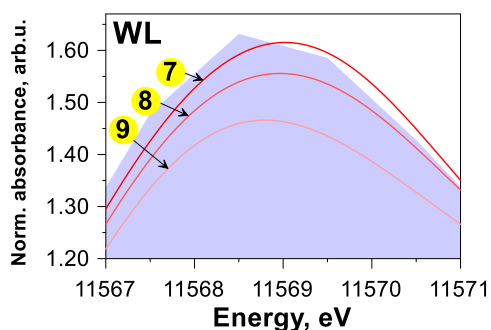
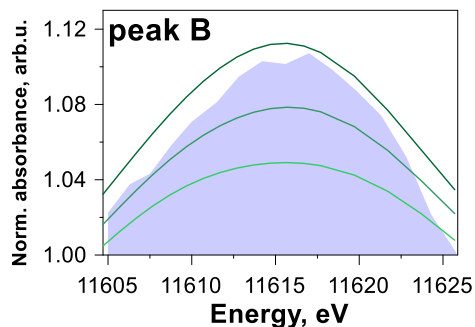
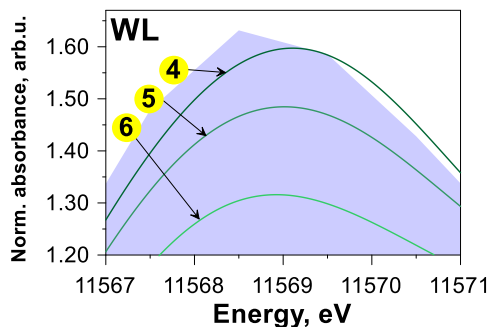
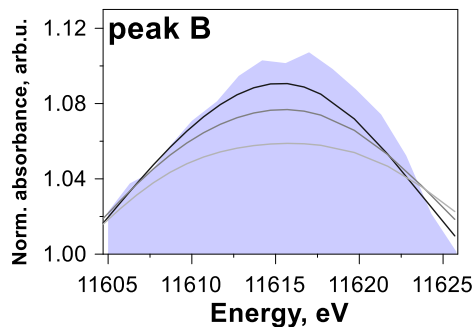
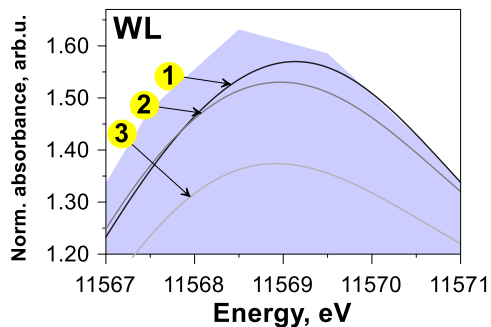


Figure A.2. The Pt L_3 -edge EXAFS spectrum of $\text{PtS}_{(\text{cr.})}$ and fit results: Fourier Transform (FT) of the k^2 -weighted EXAFS spectrum (not corrected for phase shift), *insert*: k^2 -weighted background-subtracted EXAFS spectrum. Thin black line – experiment, dotted red line – fit results. The 1st coordination sphere of Pt consists of 4 S at $R_{\text{Pt-S}} = 2.31 \pm 0.01$ Å, the 2nd coordination sphere consists of 4 and 8 Pt atoms at $R_{\text{Pt-Pt}} = 3.49 \pm 0.05$ and 3.87 ± 0.07 Å, respectively, the 3rd coordination sphere consists of 8 S atoms at $R_{\text{Pt-S}} = 4.09 \pm 0.03$ Å.

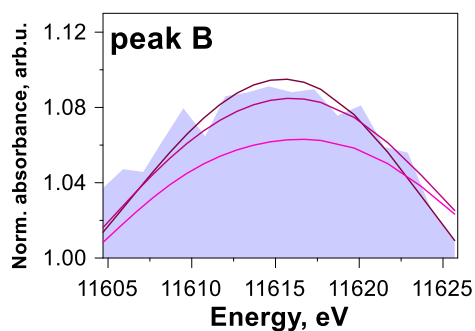
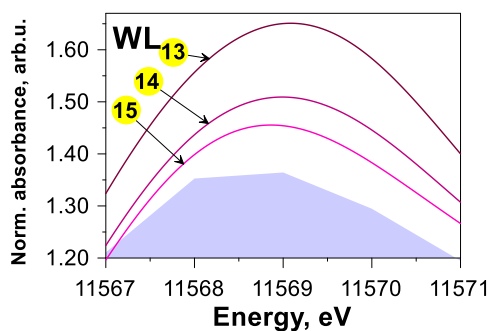
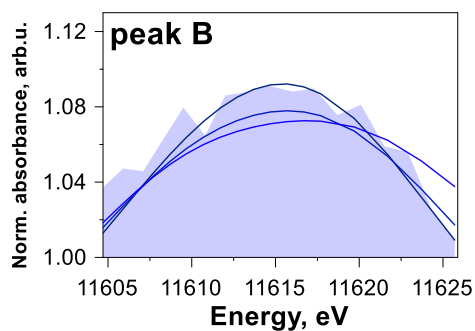
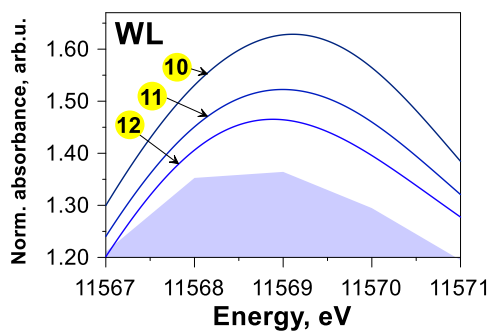
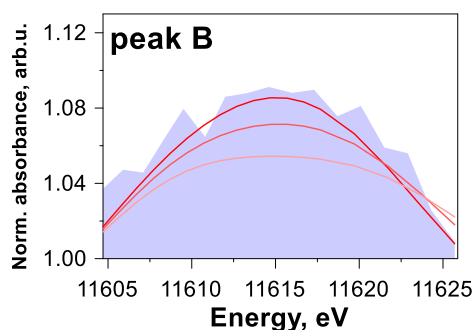
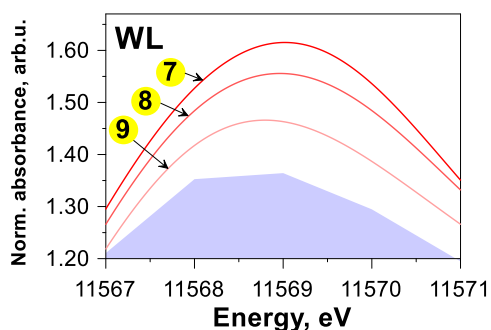
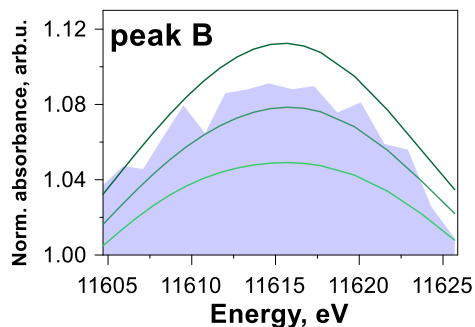
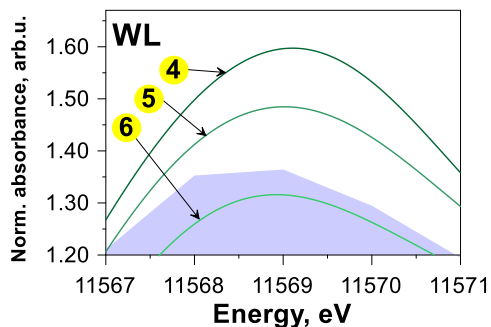
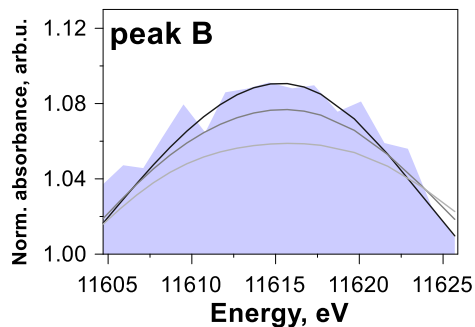
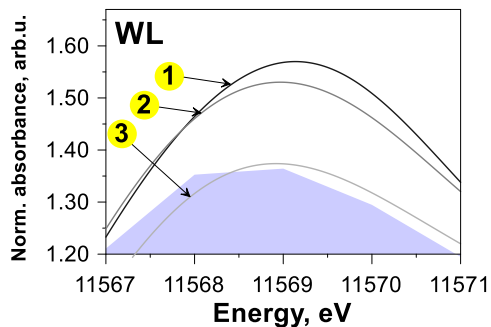
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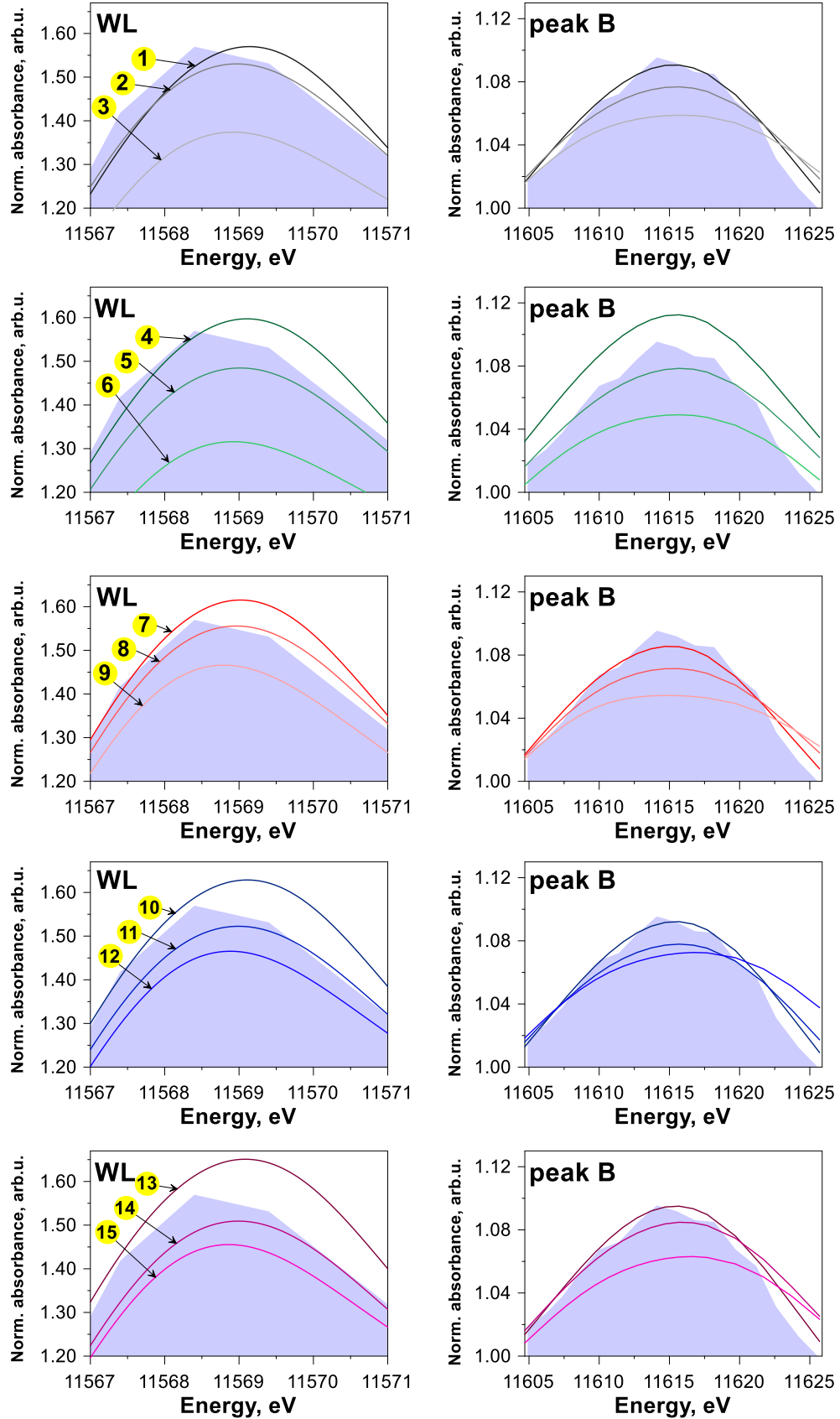
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No. 5590

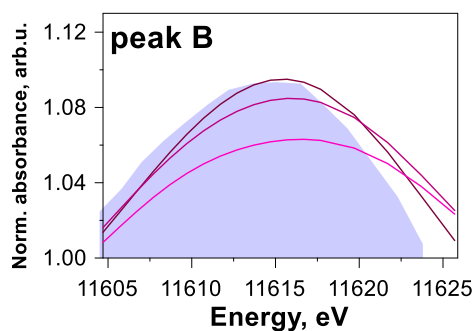
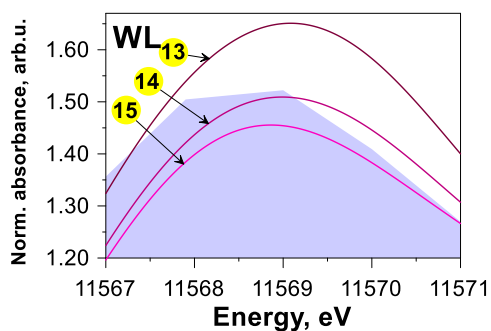
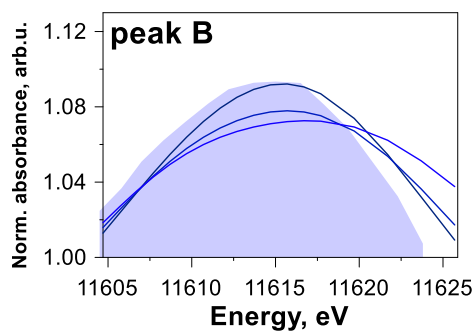
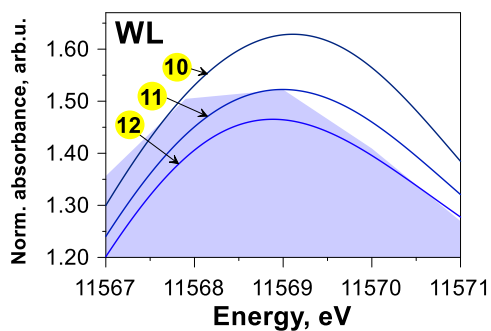
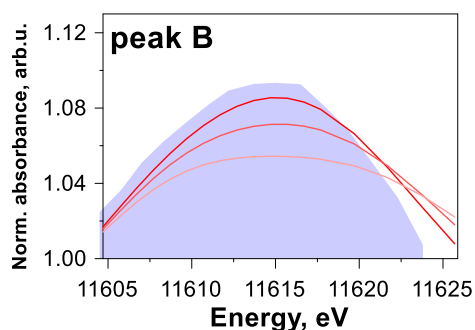
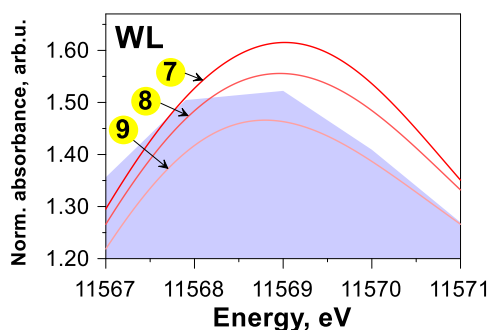
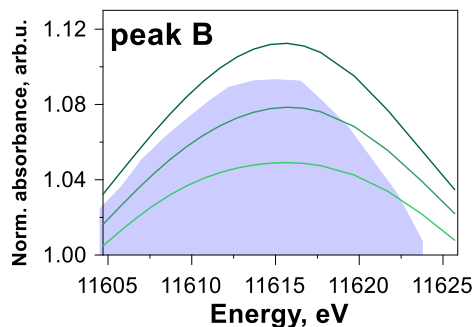
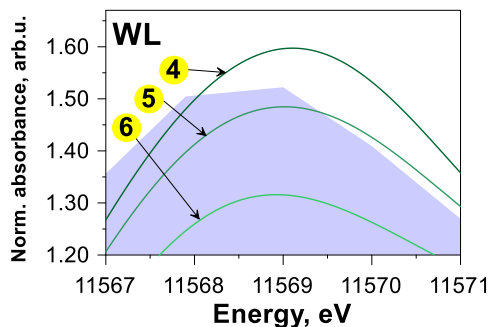
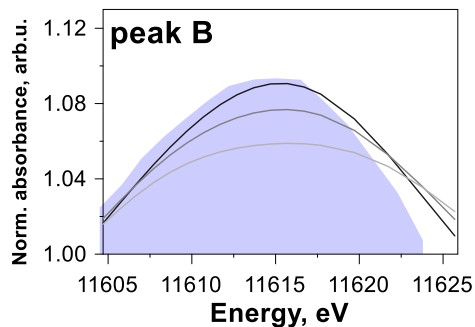
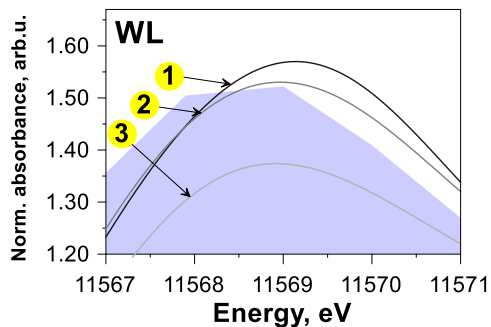


Figure A.3. Enlarged portions of Pt L_3 -edge XANES spectra of pyrrhotite samples: WL (*left*) and peak B (*right*) regions. The experimental spectra are shown by blue fields, the spectra calculated by FDMNES code are shown by solid lines. The number of S atoms in the 1st coordination shell varied from 6 to 4, the number of Fe atoms in the 2nd coordination shell varied from 5 to 8. Each model spectrum was calculated for various numbers of S and Fe atoms in the 1st and 2nd coordination shells. The distant coordination shells correspond to a pure pyrrhotite structure adopted from Wyckoff (1963). The sequence number of the model is indicated in the yellow circle in every picture and explained below:

- (1) 6 S, 8 Fe,
- (2) 5 S, 8 Fe,
- (3) 4 S, 8 Fe,
- (4) 6 S, 7 Fe (one vacancy in the Fe site at $R_{\text{Pt-Fe}} = 3.43 \text{ \AA}$),
- (5) 5 S, 7 Fe (one vacancy in the Fe site at $R_{\text{Pt-Fe}} = 3.43 \text{ \AA}$),
- (6) 4 S, 7 Fe (one vacancy in the Fe site at $R_{\text{Pt-Fe}} = 3.43 \text{ \AA}$),
- (7) 6 S, 7 Fe (one vacancy in the Fe site at $R_{\text{Pt-Fe}} = 2.84 \text{ \AA}$),
- (8) 5 S, 7 Fe (one vacancy in the Fe site at $R_{\text{Pt-Fe}} = 2.84 \text{ \AA}$),
- (9) 4 S, 7 Fe (one vacancy in the Fe site at $R_{\text{Pt-Fe}} = 2.84 \text{ \AA}$),
- (10) 6 S, 6 Fe (two vacancies in Fe sites at $R_{\text{Pt-Fe}} = 3.43 \text{ \AA}$: atoms Nos. 1, 2 in Fig. 7),
- (11) 5 S, 6 Fe (two vacancies in the Fe sites at $R_{\text{Pt-Fe}} = 3.43 \text{ \AA}$: atoms Nos. 1, 2 in Fig. 7),
- (12) 4 S, 6 Fe (two vacancies in the Fe sites at $R_{\text{Pt-Fe}} = 3.43 \text{ \AA}$: atoms Nos. 1, 2 in Fig. 7),
- (13) 6 S, 5 Fe (three vacancies in the Fe sites at $R_{\text{Pt-Fe}} = 3.43 \text{ \AA}$: atoms Nos. 1, 4, 5 in Fig. 7),
- (14) 5 S, 5 Fe (three vacancies in the Fe sites at $R_{\text{Pt-Fe}} = 3.43 \text{ \AA}$: atoms Nos. 1, 4, 5 in Fig. 7),
- (15) 4 S, 5 Fe (three vacancies in the Fe sites at $R_{\text{Pt-Fe}} = 3.43 \text{ \AA}$: atoms Nos. 1, 4, 5 in Fig. 7).