

MOLECULAR DYNAMICS SIMULATION STUDY OF PORE SIZE EFFECTS ON GAS ADSORPTION KINETICS IN ZEOLITES

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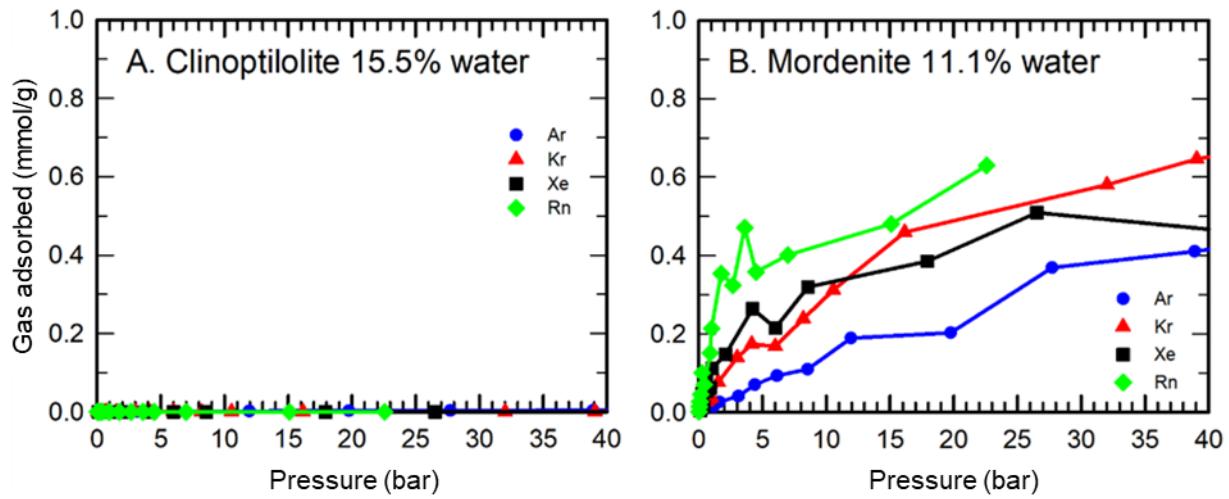


Figure S1. Adsorption isotherms at 300 K from GCMC simulations for (A) clinoptilolite and (B) mordenite in fully hydrated conditions.

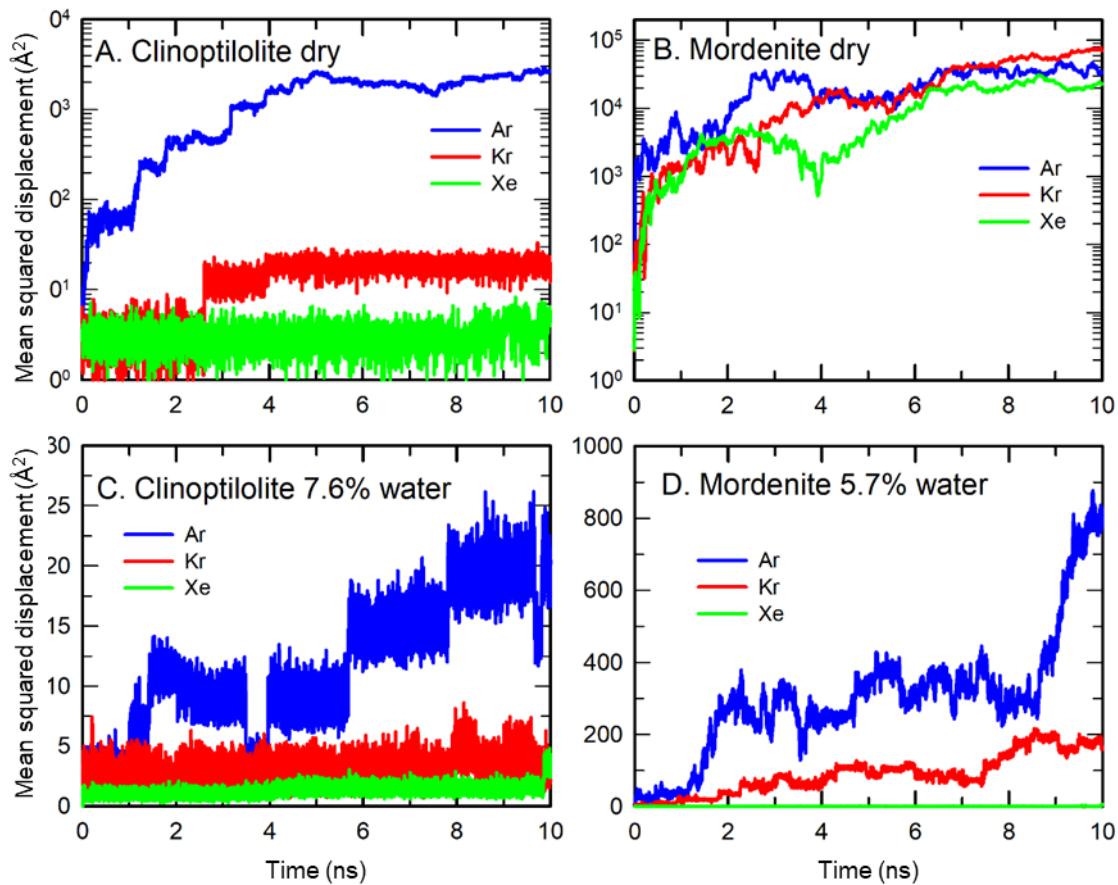


Figure S2. Time evolution of gas mean square displacement during 10 ns MD simulations at 300 K in bulk zeolites in (A, B) dry and (C, D) partially hydrated states.

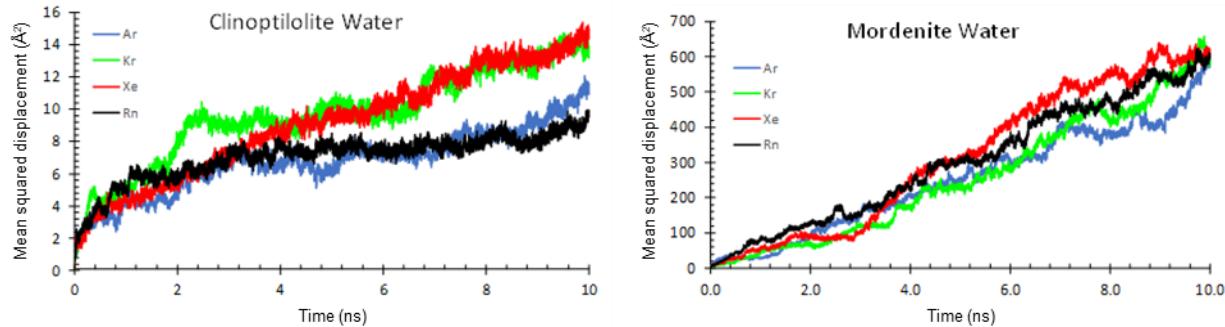


Figure S3. Time evolution of water mean squared displacement from 10-ns simulations in partially hydrated bulk zeolites with different gases. Slopes over the range 6 – 10 ns correspond to water diffusion coefficients of $0.005 \times 10^{-9} \text{ m}^2 \cdot \text{s}^{-1}$ and $0.25 \times 10^{-9} \text{ m}^2 \cdot \text{s}^{-1}$ for clinoptilolite and mordenite, respectively.

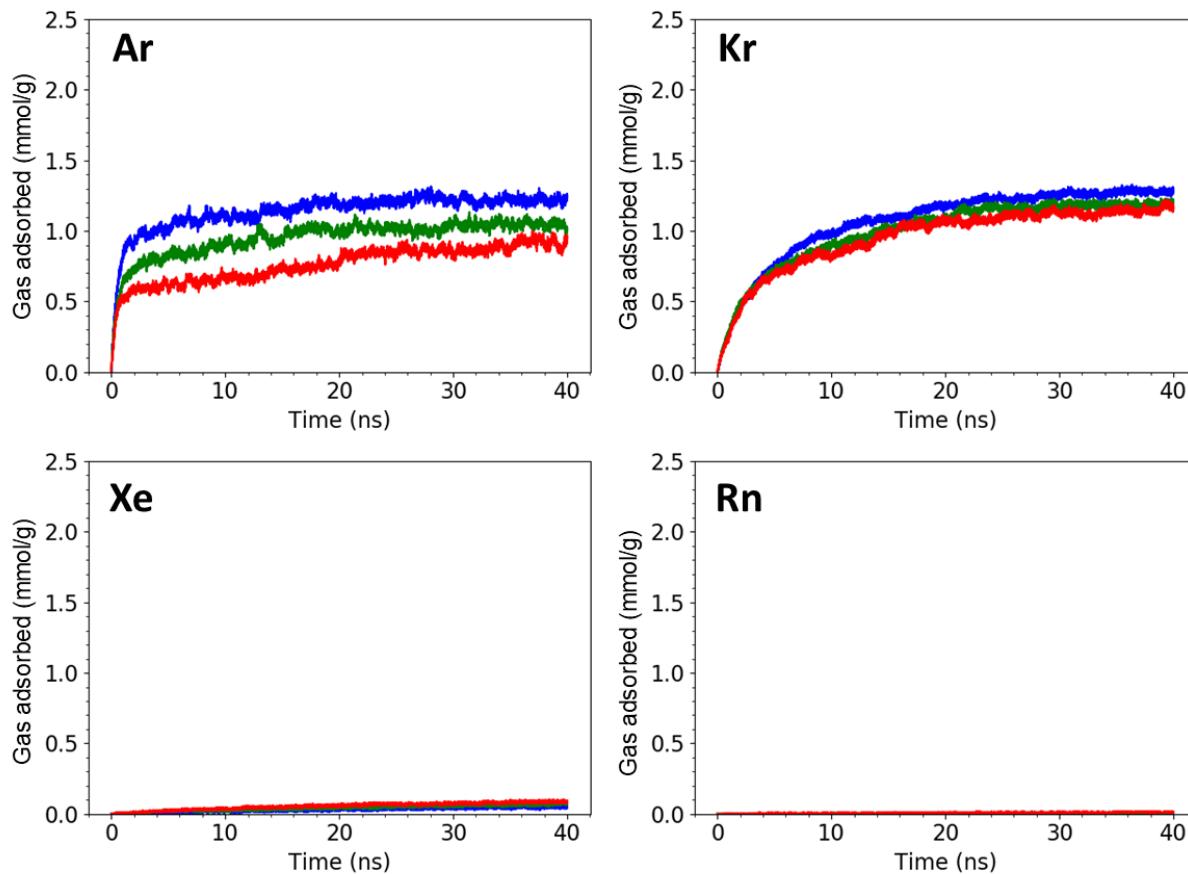


Figure S4. Results from MD simulations showing gas loading profiles in the bulk region of dry clinoptilolite slabs at low gas concentration (50 atoms on each side).

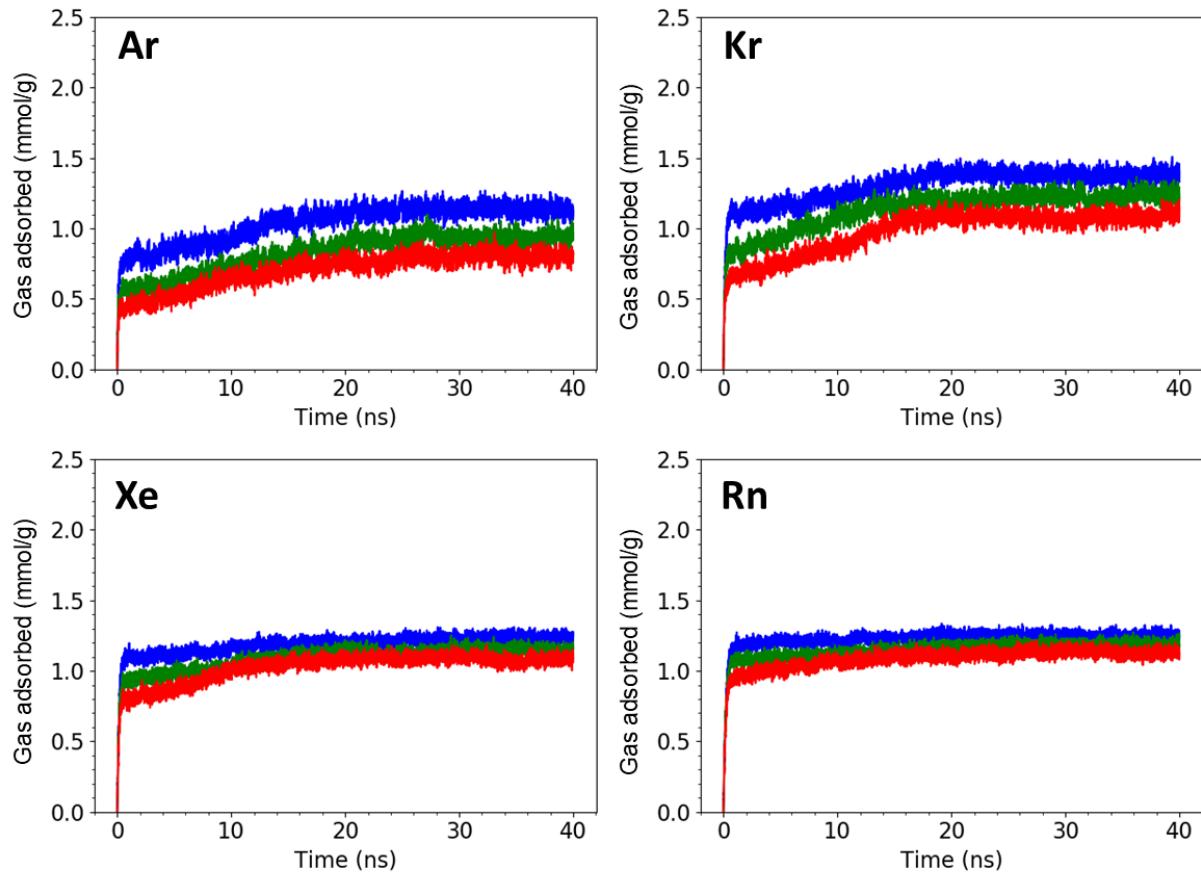


Figure S5. Results from MD simulations showing gas loading profiles in the bulk region of dry mordenite slabs at low gas concentration (50 atoms on each side).

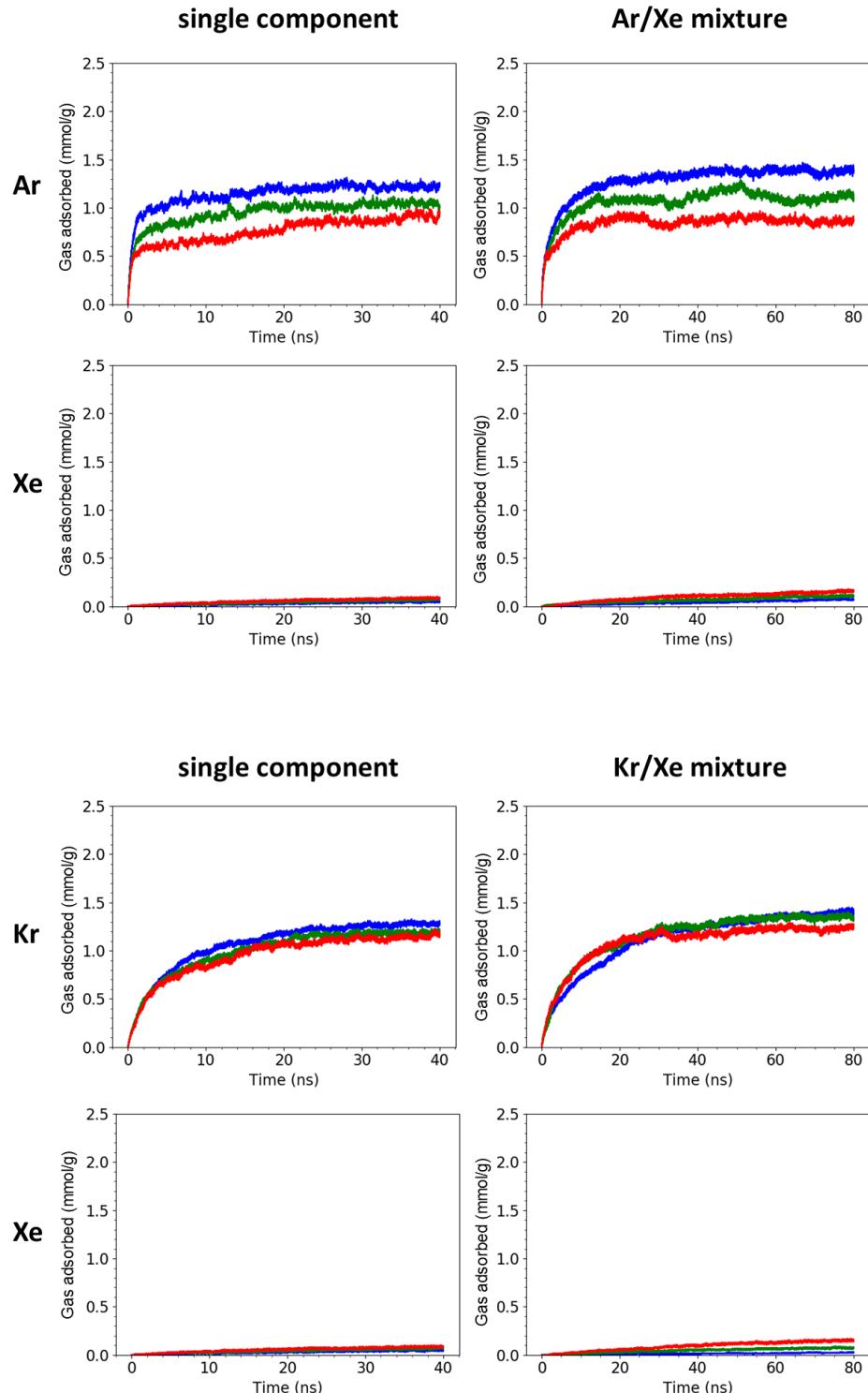


Figure S6. Results from MD simulations of Ar/Xe and Kr/Xe mixtures showing gas loading profiles in the “bulk zeolite” region of 4.4 nm clinoptilolite slabs for (dry models). Gas concentrations are 50 atoms of each gas on each side (low concentrations). Results in the left column show gas loading profiles from single gas simulations (Figure S4). Simulation temperatures are 250 K (blue), 300 K (green), and 350 K (red).

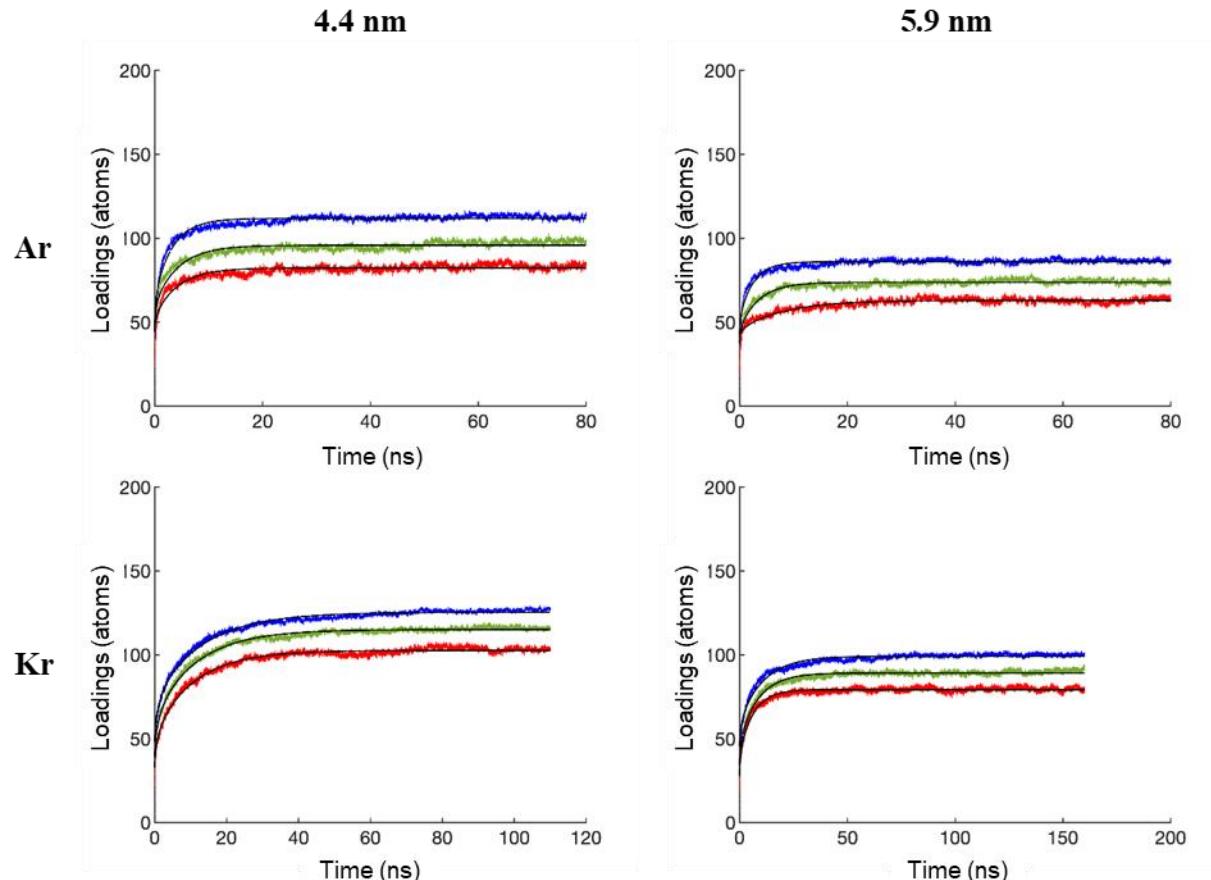


Figure S7. Kinetic model fits (black lines) of MD simulation results at 250 K (blue), 300 K (green), and 350 K (red) of total gas loading profiles (“interface” and “bulk” regions) for Ar and Kr in the 4.4 nm and 5.9 nm clinoptilolite slabs at high gas concentration (100 atoms on each side).

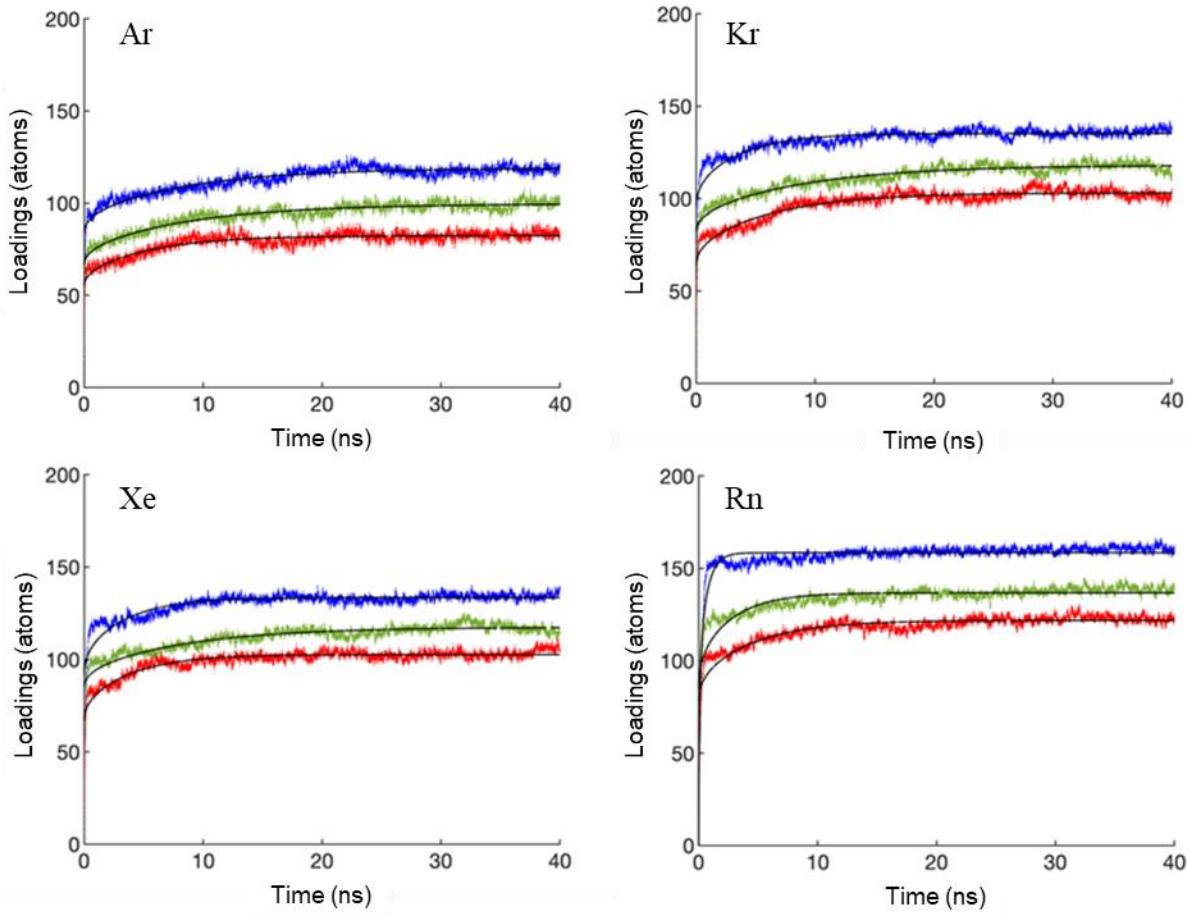


Figure S8. Kinetic model fits (black lines) of MD simulation results at 250 K (blue), 300 K (green), and 350 K (red) of total gas loading profiles (“interface” and “bulk” regions) for single component noble gases in the 3.9 nm mordenite slab at high gas concentration (100 atoms on each side).

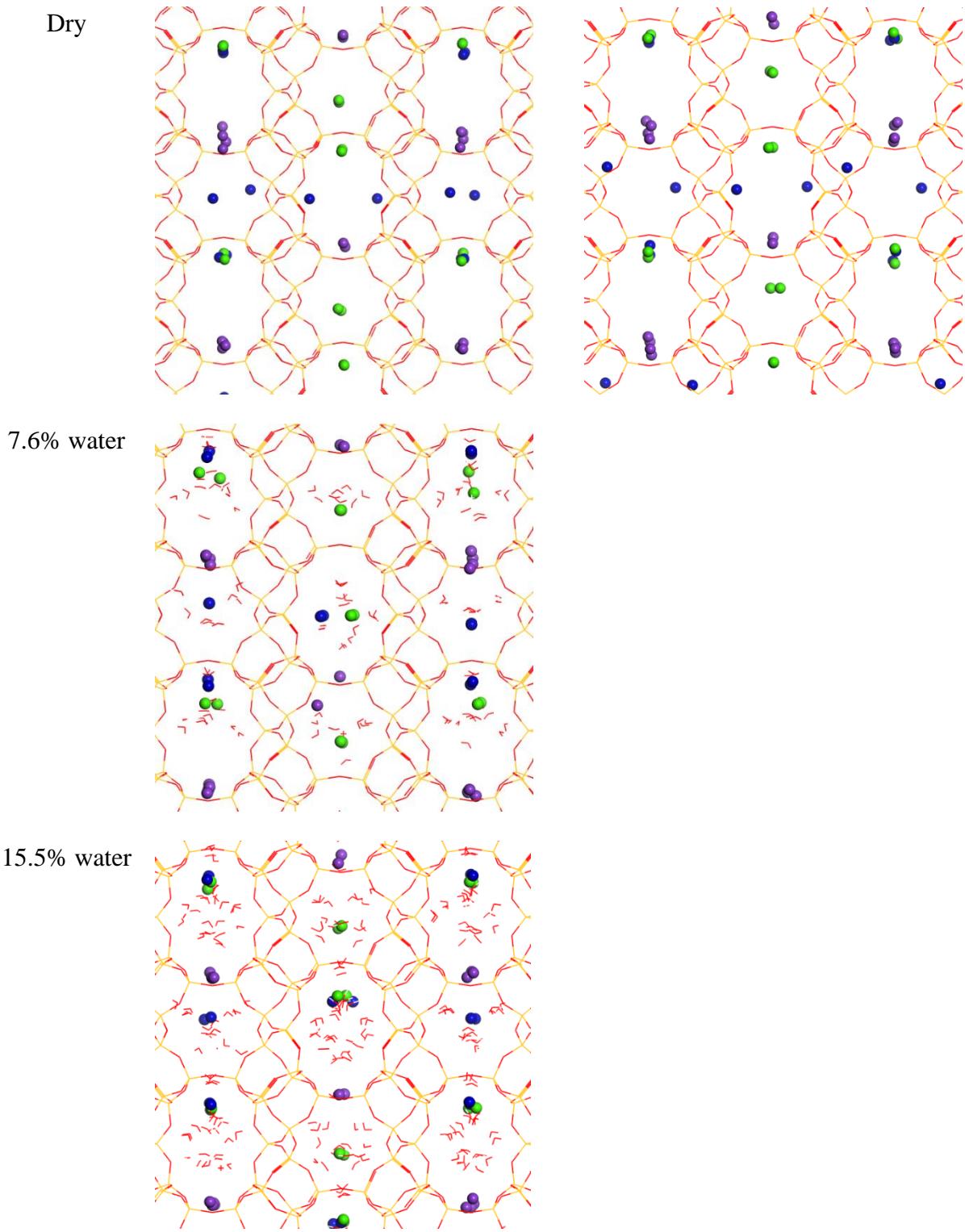


Figure S9. Snapshots from 1-ns MD simulations of clinoptilolite in dry, 7.6% water, and 15.5% water. Cations are shown as spheres (K = purple, Na = blue, Ca = green). For the dry state, snapshots are shown after simulation at 300 K (left) and after simulation at 623 K (350 °C) followed by cooling to 300 K (right).

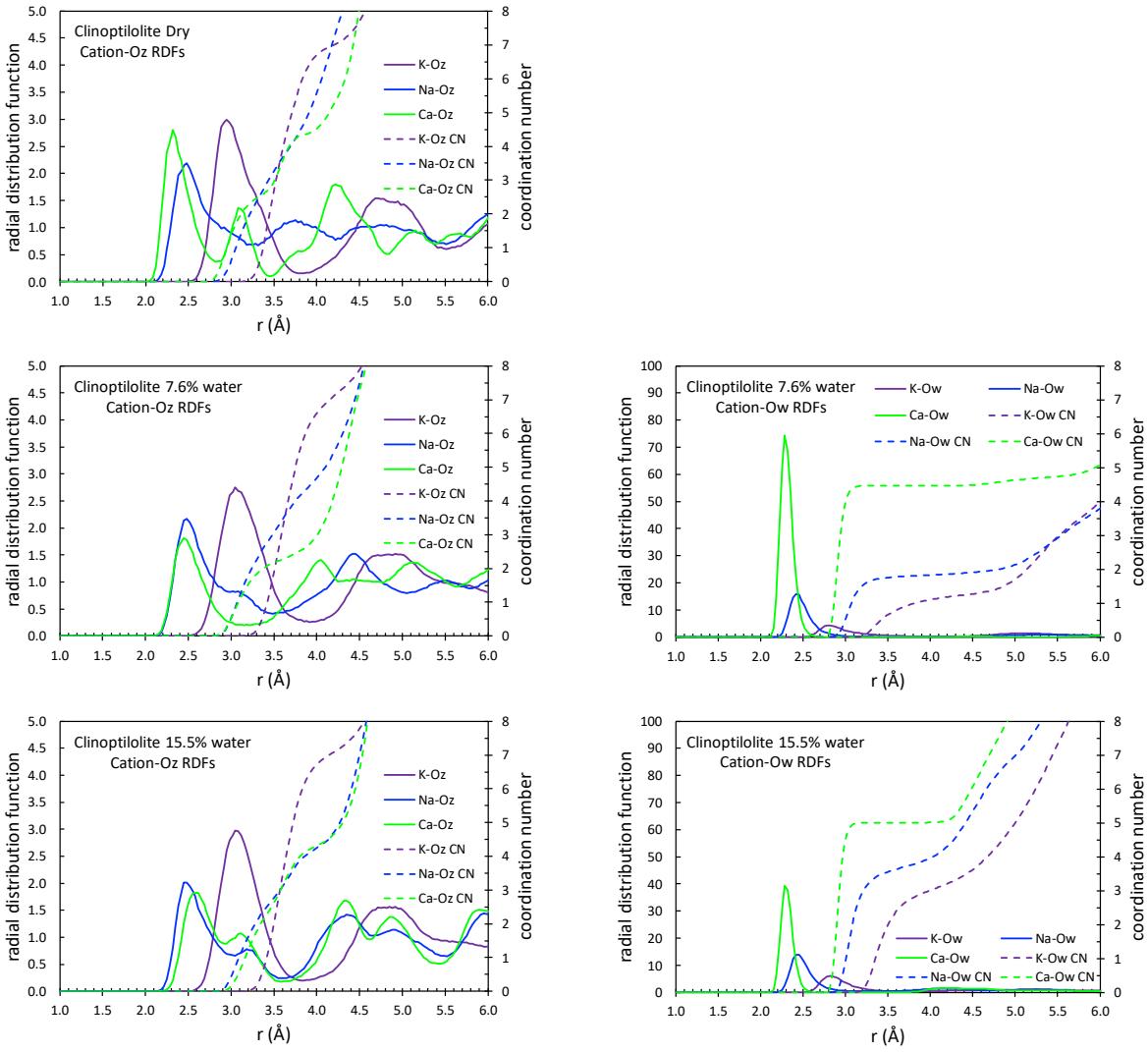


Figure S10. Cation-oxygen radial distribution functions (RDFs) and running coordination numbers (CN) from 1-ns MD simulations of clinoptilolite in dry, 7.6% water, and 15.5% water. Oz and Ow refer to zeolite oxygen and water oxygen, respectively.

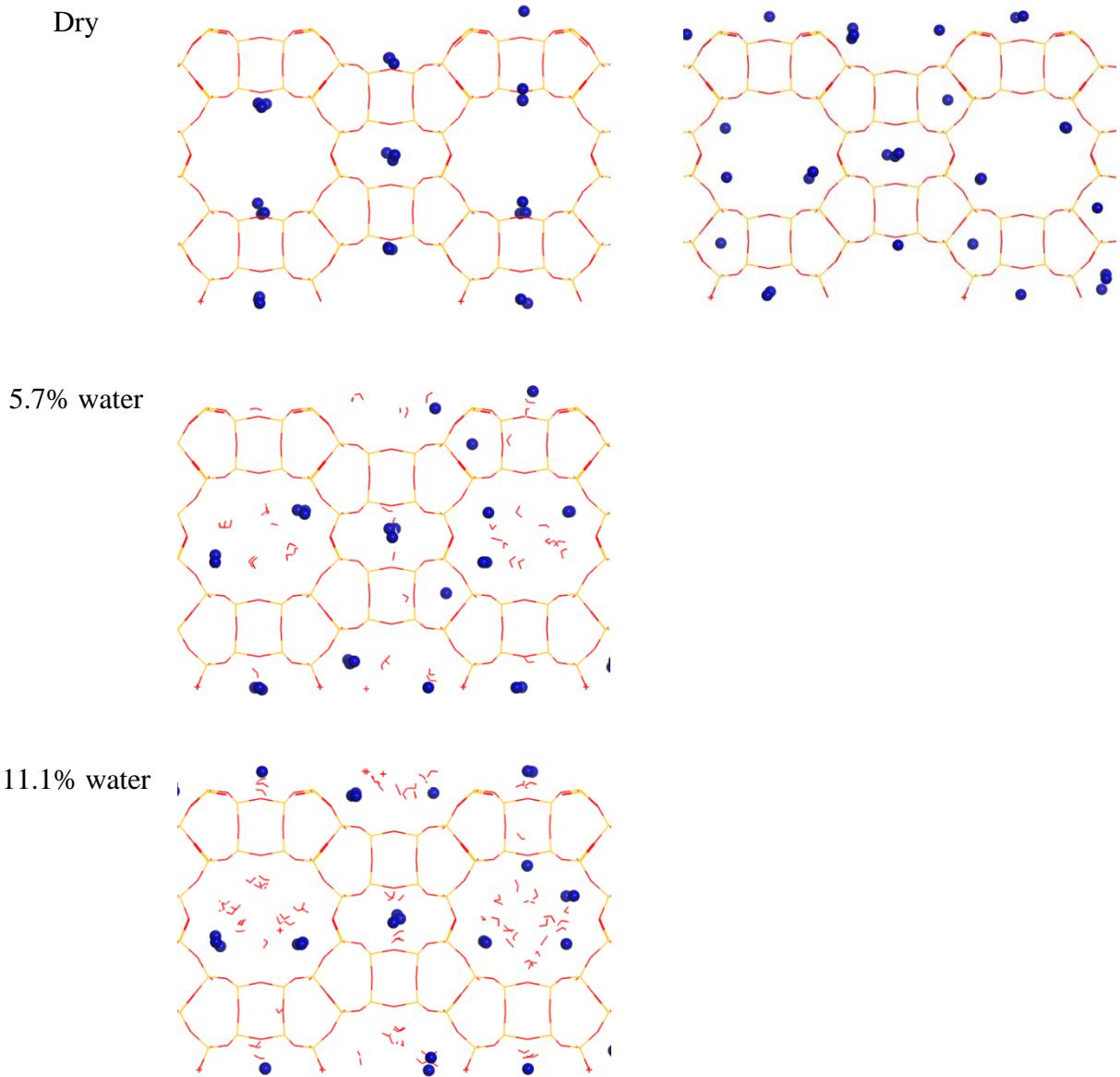


Figure S11. Snapshots from 1-ns MD simulations of mordenite in dry, 5.7% water, and 11.1% water. Cations are shown as spheres (K = purple, Na = blue, Ca = green). For the dry state, snapshots are shown after simulation at 300 K (left) and after simulation at 623 K (350 °C) followed by cooling to 300 K (right).

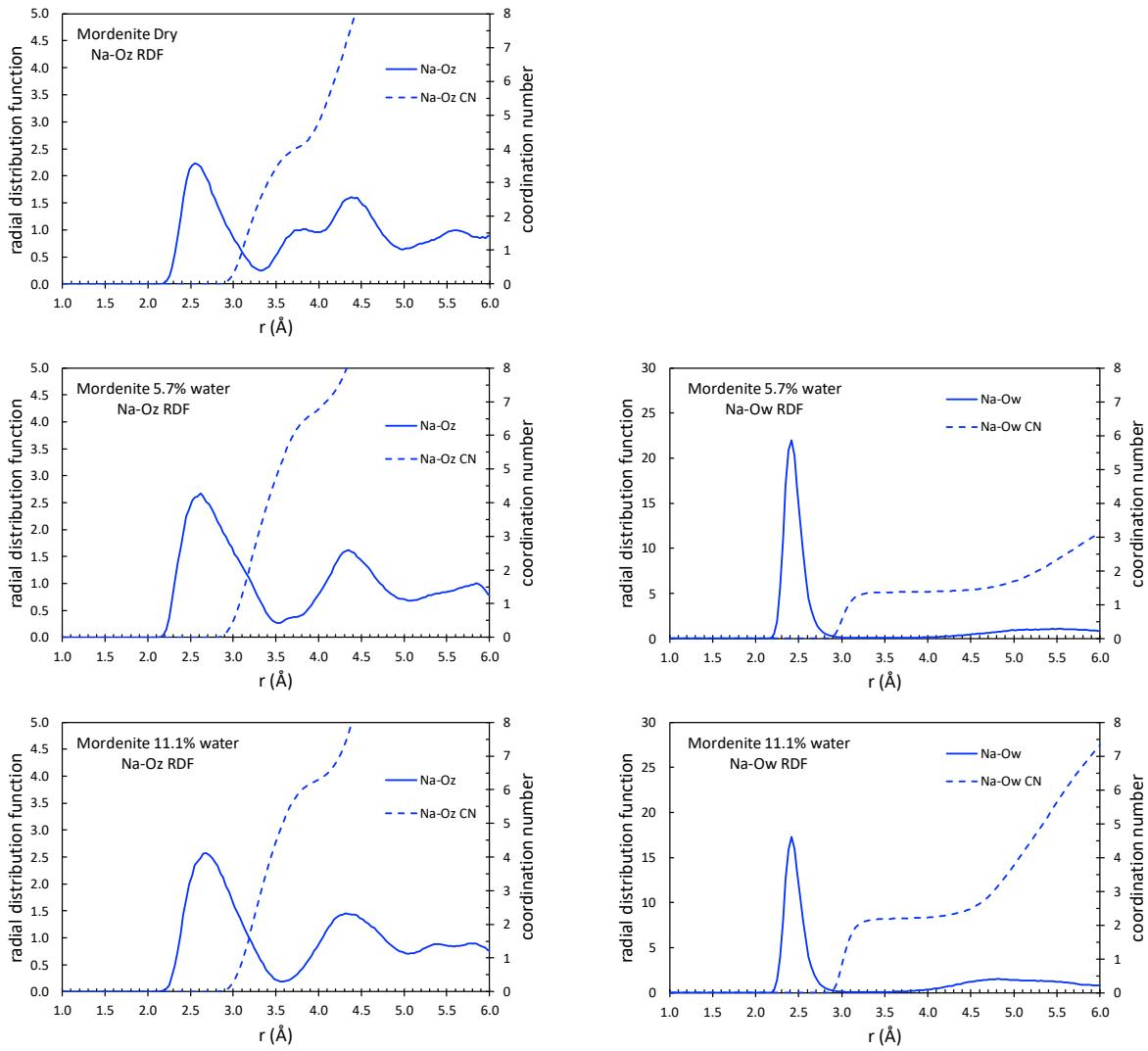


Figure S12. Sodium-oxygen radial distribution functions (RDFs) and running coordination numbers (CN) from 1-ns MD simulations of mordenite in dry, 5.7% water, and 11.1% water. Oz and Ow refer to zeolite oxygen and water oxygen, respectively.

Table S1. Gas Loadings (N_{gas} / unit cell) and Corresponding Pressures (bar) for MD Simulations of Gas Mobility in Bulk Clinoptilolite.

| Dry | | | 7.6 % H ₂ O | |
|-----|-------------------------------------|---------|-------------------------------------|---------|
| Gas | $N_{\text{gas}} / \text{unit cell}$ | P (bar) | $N_{\text{gas}} / \text{unit cell}$ | P (bar) |
| Ar | 0.5 | 1.6 | 0.5 | 6.0 |
| Kr | 0.5 | 0.4 | 0.4 | 1.2 |
| Xe | 0.5 | < 0.1 | 0.6 | 0.4 |
| Rn | 0.5 | < 0.02 | 0.4 | 0.05 |

Table S2. Gas Loadings and Corresponding Pressures for MD Simulations of Gas Mobility in Bulk Mordenite.

| Dry | | | 5.7 % H ₂ O | |
|-----|-------------------------------------|---------|-------------------------------------|---------|
| Gas | $N_{\text{gas}} / \text{unit cell}$ | P (bar) | $N_{\text{gas}} / \text{unit cell}$ | P (bar) |
| Ar | 0.8 | 2.0 | 1.2 | 12.0 |
| Kr | 1.5 | 3.0 | 1.2 | 1.2 |
| Xe | 2.3 | 0.1 | 1.8 | 0.8 |
| Rn | 1.0 | < 0.02 | 1.3 | 0.1 |

Table S3. Fitted parameters and 95% confidence intervals (parentheses) from Ar kinetic models in clinoptilolite slabs (high gas concentration).

| T(K) | <i>Slab Thickness 2w(Å)</i> | α | β (ns ⁻¹) | M_{film} (atoms) | n_∞ (atoms/ Å) | D (Å ² /ns) |
|-------------|---------------------------------|-------------------------|----------------------------------|----------------------------|--------------------------|-----------------------------|
| 250 | 43.7 | 2.587 (2.572, 2.602) | 0.1192 (0.1183, 0.1201) | 42.01 (41.75, 42.26) | 1.008 | 56.91 |
| 250 | 58.6 | 1.339 (1.333, 1.345) | 0.07827 (0.07784, 0.07871) | 45.94 (45.63, 46.24) | 1.124 | 67.19 |
| 250 | 73.8 | 0.7585 (0.7552, 0.7618) | 0.102 (0.1015, 0.1025) | 36.53 (36.13, 36.94) | 1.260 | 138.88 |
| 300 | 43.7 | 3.491 (3.472, 3.51) | 0.0901 (0.08944, 0.09076) | 37.6 (37.4, 37.8) | 0.827 | 43.02 |
| 300 | 58.6 | 2.221 (2.207, 2.234) | 0.06771 (0.06716, 0.06826) | 48.83 (48.54, 49.12) | 0.801 | 58.13 |
| 300 | 73.8 | 1.345 (1.338, 1.352) | 0.07113 (0.07069, 0.07158) | 49.16 (48.83, 49.49) | 0.872 | 96.85 |
| 350 | 43.7 | 6.171 (6.135, 6.207) | 0.04297 (0.04257, 0.04336) | 40.79 (40.66, 40.92) | 0.508 | 20.51 |
| 350 | 58.6 | 3.115 (3.094, 3.135) | 0.0732 (0.07254, 0.07386) | 44.52 (44.27, 44.77) | 0.645 | 62.84 |
| 350 | 73.8 | 2.205 (2.191, 2.219) | 0.06302 (0.06247, 0.06356) | 49.4 (49.09, 49.71) | 0.637 | 85.81 |

Table S4. Fitted parameters and 95% confidence intervals (parentheses) from Kr kinetic models used to fit simulated gas loadings in clinoptilolite slabs (high gas concentration).

| T(K) | <i>Slab Thickness</i> $2w(\text{\AA})$ | α | $\beta (\text{ns}^{-1})$ | M_{film} (atoms) | n_∞ (atoms/\text{\AA}) | D (\text{\AA}^2/ns) |
|-------------|---|-------------------------------|----------------------------------|----------------------------|----------------------------------|--------------------------|
| 250 | 4.4 | 1.869 (1.864, 1.873) | 0.0259 (0.02582, 0.02598) | 45.24 (45.12, 45.37) | 1.234 | 12.37 |
| 250 | 5.9 | 0.9649 (0.9633, 0.9665) | 0.01726 (0.01722, 0.01731) | 48.09 (47.96, 48.23) | 1.319 | 14.82 |
| 250 | 7.4 | 0.6007 (0.5998, 0.6016) | 0.01483 (0.0148, 0.01486) | 48.68 (48.54, 48.81) | 1.281 | 20.19 |
| 300 | 4.4 | 2.011 (2.007, 2.015) | 0.03285 (0.03276, 0.03293) | 34.09 (33.98, 34.2) | 1.261 | 15.68 |
| 300 | 5.9 | 1.12 (1.118, 1.121) | 0.02082 (0.02078, 0.02087) | 39.18 (39.06, 39.3) | 1.295 | 17.87 |
| 300 | 7.4 | 0.7084 (0.7072, 0.7096) | 0.01934 (0.0193, 0.01939) | 39.07 (38.9, 39.23) | 1.276 | 26.33 |
| 350 | 4.4 | 2.347 (2.341, 2.353) | 0.04688 (0.04673, 0.04703) | 27.82 (27.69, 27.94) | 1.177 | 22.38 |
| 350 | 5.9 | 1.39 (1.387, 1.392) | 0.02347 (0.02341, 0.02353) | 32.58 (32.45, 32.71) | 1.195 | 20.15 |
| 350 | 7.4 | 0.9198 (0.918, 0.9215) | 0.02242 (0.02236, 0.02247) | 35.24 (35.08, 35.41) | 1.163 | 30.53 |

Table S5. Characteristic timescales γ (μs^{-1}) from kinetic models used to fit simulated gas loadings in clinoptilolite slabs (high gas concentration).

| Gas | T (K) | 4.4 nm | 5.9 nm | 7.4 nm |
|------------|--------------|---------------|---------------|---------------|
| Ar | 250 | 17.8 | 43.7 | 177.3 |
| | 300 | 7.39 | 13.73 | 39.32 |
| | 350 | 1.13 | 7.54 | 12.96 |
| Kr | 250 | 7.41 | 18.54 | 41.10 |
| | 300 | 8.12 | 16.60 | 38.60 |
| | 350 | 8.51 | 12.15 | 26.50 |
| Xe | 250 | 0.006 | 0.039 | 0.108 |
| | 300 | 0.016 | 0.054 | 0.123 |
| | 350 | 0.039 | 0.123 | 0.249 |
| Rn | 250 | 0.338 | 0.098 | 0.403 |
| | 300 | 0.236 | 0.089 | 0.341 |
| | 350 | 0.232 | 0.079 | 0.182 |

Atom types, masses, and charges for clinoptilolite in LAMMPS data file format

units real

Masses

| | | | |
|----|-----------|------|-------------------|
| 1 | 28.086 | # sz | zeolite Si |
| 2 | 15.9994 | # oz | zeolite O |
| 3 | 15.9994 | # o* | SPC water O |
| 4 | 39.102 | # K | |
| 5 | 1.00797 | # h* | SPC water H |
| 6 | 22.99 | # Na | |
| 7 | 40.0798 | # Ca | |
| 8 | 39.95 | # Ar | |
| 9 | 83.80 | # Kr | |
| 10 | 131.293 | # Xe | |
| 11 | 222.0 | # Rn | |
| 12 | 15.999400 | # oh | zeolite silanol O |
| 13 | 1.007970 | # ho | zeolite silanol H |

Atoms # full

| | | | | | | | | | | |
|----|---|----|-----------|--------------|--------------|-------------|----|---|---|------|
| 1 | 1 | 1 | 1.299800 | 6.703092071 | 2.825806651 | 3.058940500 | 0 | 0 | 0 | # sz |
| 2 | 1 | 2 | -0.733300 | 7.186179122 | 1.281011288 | 2.895677400 | 0 | 0 | 0 | # oz |
| 3 | 1 | 3 | -0.820000 | 0.443236726 | 14.534976412 | 7.517279000 | 0 | 0 | 0 | # o* |
| 4 | 1 | 4 | 1.000000 | 5.749371076 | 3.702454326 | 8.970500000 | 0 | 0 | 0 | # K |
| 5 | 1 | 5 | 0.410000 | 14.487733424 | 14.256207043 | 8.100986293 | -1 | 0 | 0 | # h* |
| 6 | 1 | 6 | 1.000000 | 1.286945185 | 0.638136326 | 8.970500000 | 0 | 0 | 0 | # Na |
| 7 | 1 | 7 | 2.000000 | 2.194336622 | 7.259788262 | 0.019999000 | 0 | 0 | 0 | # Ca |
| 8 | 1 | 8 | 0.000000 | 28.56 | 1.06 | 8.43 | 0 | 0 | 0 | # Ar |
| 9 | 1 | 9 | 0.000000 | 21.453 | 16.185 | 8.507 | 0 | 0 | 0 | # Kr |
| 10 | 1 | 10 | 0.000000 | 12.383 | 7.145 | 0.054 | 0 | 0 | 0 | # Xe |
| 11 | 1 | 11 | 0.000000 | 25.509 | 0.217 | 0.422 | 0 | 0 | 0 | # Rn |
| 12 | 1 | 12 | -0.777200 | 51.823 | 12.912 | 15.102 | 0 | 0 | 0 | # oh |
| 13 | 1 | 13 | 0.410000 | 52.280 | 13.126 | 15.942 | 0 | 0 | 0 | # ho |

Atom types, masses, and charges for mordenite in LAMMPS data file format

units real

Masses

```
1 28.086 # sz zeolite Si
2 15.9994 # oz zeolite O
3 15.9994 # o* SPC water O
4 22.99 # Na
5 1.00797 # h* SPC water H
6 15.999400 # oh zeolite silanol O
7 1.007970 # ho zeolite silanol H
```

Atoms # full

```
1 1 1 1.300000 3.553676000 8.756489000 4.031845600 0 0 0 # sz
2 1 2 -0.712500 2.248244000 8.571926000 3.129193600 0 0 0 # oz
3 1 3 -0.820000 18.111000000 14.826561000 2.271674200 0 0 0 # o*
4 1 4 1.000000 36.475945800 10.103798900 3.839279840 -1 0 0 # Na
5 1 5 0.410000 1.497493374 -0.080279899 3.402823501 0 1 0 # h*
6 1 3 -0.767600 15.808864525 8.422441465 50.642987944 0 0 0 # oh
7 1 4 0.410000 15.918548187 9.393552874 50.779020654 0 0 0 # ho
```

van der Waals parameters in LAMMPS format

```
pair_coeff 2 4 buck/coul/long 121418 0.2833 3577 # oz K
pair_coeff 2 6 buck/coul/long 121418 0.2469 1501 # oz Na
pair_coeff 2 7 buck/coul/long 121418 0.2469 1501 # oz Ca
pair_coeff 3 3 lj/cut/coul/long 0.1553 3.1690 # o* o*
pair_coeff 4 4 lj/cut/coul/long 0.1232 3.3320 # K K
pair_coeff 6 6 lj/cut/coul/long 0.0999 2.5840 # Na Na
pair_coeff 7 7 lj/cut/coul/long 0.0999 2.5840 # Ca Ca
pair_coeff 8 8 lj/cut/coul/long 0.2385 3.4050 # Ar Ar
pair_coeff 9 9 lj/cut/coul/long 0.3378 3.6900 # Kr Kr
pair_coeff 10 10 lj/cut/coul/long 0.4193 4.1000 # Xe Xe
pair_coeff 11 11 lj/cut/coul/long 0.5962 4.1700 # Rn Rn
```

bond stretch and angle bend parameters in LAMMPS format

```
units      real
bond_style harmonic
angle_style harmonic
bond_coeff 1  1383.1  1.65 # sz-oz
bond_coeff 2  1383.1  1.65 # sz-oh
bond_coeff 3  553.935  1.0 # oh-ho
bond_coeff 4  553.935  1.0 # o*-h*
angle_coeff 1  298.1   110  # oz-sz-oz
angle_coeff 2  59.6    145  # sz-oz-sz
angle_coeff 3  298.1   110  # oz-sz-oh
angle_coeff 4  24.5    118.5 # sz-oh-ho
angle_coeff 5  45.77   109.47 # h*-o*-h*
```